

Deacon's Challenge

No. 13 Answer

Your on-call laboratory service uses 30 different methods, each of which has a 1% probability of failing QC criteria during the course of a night. Assuming that QC of any method is independent of that of the other methods, what is the probability that on any one night all methods will pass the QC criteria?

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The probability of a channel failing QC is $1\% = 0.01$

There are only two possible outcomes: pass or fail.

Therefore the probability of a channel passing QC is $1 - 0.01 = 0.99$

This problem is analogous to flipping a coin. The joint probability of two independent events is the product of their individual probabilities.

Thus if a coin is tossed once the probability of 'heads' is 0.5. If the coin is tossed again then the probability of it landing 'heads' on both occasions is $0.5 \times 0.5 = 0.25$. Similarly if the probability of one channel passing QC is 0.99, then the probability of two channels passing is $0.99 \times 0.99 = 0.98$. The chance of three different channels passing is given by $0.99 \times 0.99 \times 0.99 = 0.97$ i.e. $(0.99)^3$.

The general rule is:

Probability of event occurring on n occasions = (probability of event occurring on a single occasion) ^{n}

Therefore the probability of 30 channels passing QC = $(0.99)^{30} = 0.74$

If your calculator does not have the facility to calculate x^y then the result can be easily calculated using logs:

$$\begin{aligned}\log_{10} (\text{probability of 30 channels passing}) &= 30 \times \log_{10} 0.99 \\ &= 30 \times 0.00436 = -0.131\end{aligned}$$

$$\text{Probability of 30 channels passing} = \text{antilog} (-0.131) = 0.74$$

Question No. 14

A 0.5 mL sample of urine is extracted into dichloromethane. An aliquot of the extract is analysed by HPLC and found to give an apparent original concentration of 320 nmol/L of analyte Y. 100 mL of Y standard with a concentration of 880 nmol/L is added to a further 0.5 mL sample of the same urine and the sample mixed. 0.5 mL of the mixed sample is then processed as before, giving a measured concentration of 405 nmol/L. Calculate the recovery of analyte Y.

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