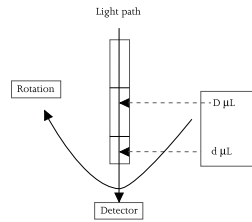


Deacon's Challenge

No. 25 Answer

A centrifugal analyser is designed so that the light travels on a longitudinal path through the rotating cuvette (which has a constant cross-section $C \text{ cm}^2$) rather than perpendicularly through the sides of the cuvette as is more usual. A solution of a light absorbing compound Y, volume $d \text{ }\mu\text{L}$ at a concentration of $y \text{ mmol/L}$, is diluted with a volume $D \text{ }\mu\text{L}$ of an optically clear reagent.



Using the Beer-Lambert equation, prove that the absorbance of light through the diluted solution of Y is independent of the volume of diluent (D) when absorbance is measured longitudinally in this system.

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The Beer-Lambert equation is:

	A	$=$	$\epsilon \times l \times c$
Where	A	$=$	absorbance
	ϵ	$=$	molar absorptivity
	l	$=$	path length
	c	$=$	molar concentration

$$\text{Volume of cuvette} = \text{Cross sectional area} \times \text{path length} \quad (i)$$

Where	C	$=$	Cross sectional area (cm^2)
	l	$=$	Depth of solution in cuvette (cm)
	$(D + d)$	$=$	volume of cuvette (μL)

$$\text{Since } 1000 \mu\text{L} = 1 \text{ cm}^3, \quad \frac{(D + d)}{1000} = \text{volume of cuvette (cm}^3\text{)}$$

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Substitute these values into equation (i) and rearrange to give an expression for l in terms of D , d and C :

$$\frac{(D + d)}{1000} = C \times l$$

$$l = \frac{(D + d)}{1000C}$$

Calculate the concentration of Y in the cuvette:

$$\begin{aligned} \text{Concn in cuvette (mmol/L)} &= \frac{\text{initial conc (y mmol/L)} \times \text{Vol of Y (d L)}}{\text{Final vol (D + d) L}} \\ &= \frac{(y \times d)}{(D + d)} \end{aligned}$$

Substitute these derived values for path length (l) and final concn of X into the Beer-Lambert equation:

$$A = \epsilon \times \frac{(D + d)}{1000C} \times \frac{(y \times d)}{(D + d)}$$

Cancelling the $(D + d)$ terms which appear on both the top and bottom of the equation eliminates the D term:

$$A = \frac{\epsilon \times y \times d}{1000C}$$

Therefore the absorbance is independent of the volume of diluent (D). ■

Question No. 26

A urine collection was handed in by a patient which he said he had collected over the previous day. Calculate the creatinine clearance given that the sample was found to have a creatinine concentration of 7.2 mmol/L in a volume of 3.2 L . The serum creatinine concentration taken during the collection was $94 \mu\text{mol/L}$. Give the most likely cause for this result.

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