

Deacon's Challenge

No 172 - Answer

A patient has the following results:

Serum copper	12.5 $\mu\text{mol/L}$	(13-26)
Serum caeruloplasmin	155 mg/L	(200-450)

Assuming that caeruloplasmin is the only protein in serum to bind a significant amount of copper, calculate the serum concentration of free (unbound) copper.

Copper content of caeruloplasmin	0.3%
Atomic mass of copper	63.6

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$$\text{Bound copper (mg/L)} = \frac{\text{caeruloplasmin (mg/L)} \times \% \text{ copper content}}{100}$$

Multiply by 1,000 to convert from mg/L to $\mu\text{g/L}$:

$$\text{Bound copper (}\mu\text{g/L)} = \frac{\text{caeruloplasmin (mg/L)} \times \% \text{ copper content} \times 1,000}{100}$$

To convert from $\mu\text{g/L}$ to $\mu\text{mol/L}$ divide by the atomic mass of copper:

$$\text{Bound copper (}\mu\text{mol/L)} = \frac{\text{caeruloplasmin (mg/L)} \times \% \text{ copper content} \times 1,000}{100 \times \text{Atomic mass of copper}}$$

Substitute:

Caeruloplasmin = 155 mg/L ; % copper content = 0.3%; atomic mass of copper = 63.6

$$\text{Bound copper} = \frac{155 \times 0.3 \times 1,000}{100 \times 63.6} = 7.3 \mu\text{mol/L (2 sig figs)}$$

$$\text{Total copper (}\mu\text{mol/L)} = \text{Bound copper (}\mu\text{mol/L)} + \text{Free copper (}\mu\text{mol/L)}$$

Substitute: Total copper = 12.5 $\mu\text{mol/L}$ and bound copper = 7.3 $\mu\text{mol/L}$

$$12.5 = 7.3 + \text{Free copper (}\mu\text{mol/L)}$$

$$\text{Free copper} = 12.5 - 7.3 = 5.2 \mu\text{mol/L}$$

Question 173

A screening programme for Down's Syndrome has a screen positive rate of 2.5% and a detection rate of 82%. Calculate the probability that a pregnancy judged to be at low risk will result in an affected child, given that the incidence of Down's Syndrome at term is 1.84/1000 live births in the absence of selective abortion. State any assumptions made.

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