

Deacon's Challenge

No. 82 Answer

A man has been arrested for suspected drink-driving shortly after arriving back home. He claims to have drunk two half-litre bottles of beer in the hour previously to driving, and on returning home had drunk a double brandy (50 mL), immediately prior to being arrested and providing a blood sample.

His plasma ethanol concentration was 1.15 g/L.

The alcoholic strength of the beer was 4.0% (alcohol by volume, ABV), and the brandy was 40% (ABV). Liquid ethanol has a density of 0.789 g/mL.

The man weighs 90 Kg; the volume of distribution for ethanol is 0.68 L/Kg.

Is the man's claim credible? State any assumptions you make in your answer.

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First convert the concentration of alcohol in beer and brandy to g/100 mL:

$$\text{Density} = \frac{\text{Weight}}{\text{Volume}} \quad \text{so that} \quad \text{Weight} = \text{Density} \times \text{Volume}$$

$$\text{and} \quad \text{ethanol (g/100 mL)} = 0.789 \times \text{ethanol (\% by vol)}$$

$$\text{Ethanol in beer} = 4.0 \text{ mL/100 mL} = 0.789 \times 4.0 = 3.16 \text{ g/100 mL (3 sig figs)}$$

$$\text{Ethanol in brandy} = 40 \text{ mL/100 mL} = 0.789 \times 40 = 31.6 \text{ g/100 mL (3 sig figs)}$$

Next calculate the total weights of ethanol consumed in each drink:

Each 100 mL beer contains 3.16 g ethanol, therefore

$$2 \times 0.5 \text{ L contains } 3.16 \times 10 = 31.6 \text{ g}$$

Since brandy contains 31.6 g/100 mL,

$$50 \text{ mL contains } \frac{31.6}{2} = 15.8 \text{ g ethanol}$$

$$\text{Claimed total ethanol consumed} = 31.6 + 15.8 = 47.4 \text{ g}$$

This amount of ethanol is contained in its total volume of distribution (V_d).

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$$\begin{aligned} \text{Total } V_d &= \text{Body wt (Kg)} \times V_d \text{ (L/Kg)} \\ &= 90 \times 0.68 = 61.2 \text{ L} \end{aligned}$$

If the total weight of ethanol consumed is distributed throughout its V_d (i.e. if totally absorbed and not metabolized or excreted) then the maximum plasma concentration is given by:

$$\frac{\text{Weight consumed (g)}}{\text{Total } V_d \text{ (L)}} = \frac{47.4}{61.2} = 0.77 \text{ g/L (2 sig figs)}$$

Since this is likely to be an overestimate (due to incomplete absorption and some metabolism and excretion) and the measured concentration (1.15 g/L) is far in excess of this, the man's claim is not credible.

Question 83

It has been suggested that a proposed analytical goal for an analyte is that the between batch analytical coefficient of variation should not exceed one half of the "true biological" inter-individual coefficient of variation. Calculate the percentage "expansion" of the measured reference range over the true biological reference range when this analytical goal is exactly met.